

## CAFFELOYLTYROSINE FROM GREEN ROBUSTA COFFEE BEANS

MICHAEL N. CLIFFORD, BRIAN KELLARD\* and ERIC AH-SING

The Department of Biochemistry, University of Surrey, Guildford, Surrey GU2 5XH, U.K.

(Received 4 November 1988)

**Key Word Index**—*Coffea canephora*; Rubiaceae; robusta coffee beans; *N*-caffeoyltyrosine.

**Abstract**—A component found in commercial green robusta coffee beans from many origins, but particularly characteristic of those from Angola, has been characterised as *N*-caffeoyltyrosine.

### INTRODUCTION

From the results of paper chromatography [1] it has been apparent since 1960 that the chlorogenic acid (CGA)-rich fraction of commercial green robusta coffee beans from Angola might contain substances not present in the corresponding fraction of robustas from other origins. It was suggested [1] that *p*-coumaric acid was one of the components peculiar to Angolan beans. Studies in this laboratory have confirmed that commercial Angolan robustas are significantly ( $p = 0.001$ ) different [2, 3], but the presence of *p*-coumaric acid could not be confirmed using reversed phase HPLC. This paper reports the isolation and characterisation of the main component contributing to the unique chromatographic profile of Angolan beans.

### RESULTS AND DISCUSSION

A concentrated extract from a green Angolan robusta was subjected to reversed phase HPLC and the fraction containing component 7 repeatedly collected, bulked and concentrated under reduced pressure. Homogeneity, checked by analytical HPLC, was *ca* 98% as judged by peak areas at 276 and 313 nm.

Failure to isomerise and form a methyl cinnamate when treated with tetramethylammoniumhydroxide [4] indicated that this component was not a typical CGA. The UV spectrum in methanol showed maxima at 272 and 313 nm, the latter being 75% as intense. Positive reactions to ninhydrin (purple-grey) and molybdate (yellow) implied the presence respectively of  $-\text{NH}_2$  or  $>\text{NH}$  groups, and a 1,2-dihydroxy or 1,2,3-trihydroxyphenyl residue. A negative response to 4-dimethylaminocinnamaldehyde indicated that component 7 was not an indole derivative such as *N*- $\beta$ -caffeoyltryptophan reported in robusta coffee beans by Morishita *et al.* [5].

Acid hydrolysis of component 7 yielded eight products, seven of which had stronger *A* at 313 than at 280 nm. The eighth component, which was detectable at 280 but not at 313 nm, co-chromatographed with authentic *L*-tyrosine in two analytical systems.

Proton NMR in  $\text{D}_2\text{O}$  at 300 MHz clearly showed component 7 to contain 12 unexchangeable protons. As listed in Table 1, these consisted of seven aromatic protons, two *trans* vinyl protons and three aliphatic protons. The two, two-proton doublets showing ortho coupling were assigned to a 1,4-disubstituted aromatic ring consistent with a tyrosine residue. The three aliphatic protons were assigned by their chemical shifts and coupling constants to the  $-\text{CH}_2-\text{CH}<$  tyrosine side chain. The downfield shift (0.78 ppm) of the  $\text{H}_\alpha$  proton relative to authentic *L*-tyrosine suggested that the associated amino function was acylated. The remaining aromatic protons, and the two *trans* vinyl protons, were assigned to a 1,3,4-trisubstituted aromatic ring. This part of the spectrum, after making allowances for the effects of acylation, was consistent with *trans*-caffeic acid (3,4-dihydroxycinnamic acid) which, unlike tyrosine, gives a yellow colour with the molybdate reagent.

By spiking the acid hydrolysate of component 7 it was possible to confirm that caffeic acid was one of the 313 nm-absorbing components present. However, after hydrolysis for only 1 hr the caffeic acid yield was considerably less relative to tyrosine than the 1:1 ratio indicated by NMR. Accordingly an equimolar mixture of caffeic acid and *L*-tyrosine was subjected to the hydrolysis procedure, and found to produce chromatograms identical to those given by the hydrolysate of component 7.

Taken collectively, these observations strongly suggest that component 7 is *N*-*trans*-caffeoyl-*L*-tyrosine which does not seem to have been reported previously. According to Meilgaard and Ravn [6] *N*-caffeoyltyramine occurs widely, *N*-caffeoyl DOPA occurs in Fabaceae, and as mentioned above, *N*- $\beta$ -caffeoyl-*L*-tryptophan has been reported in robusta coffee beans.

### EXPERIMENTAL

**Materials.** Commercial green robusta coffee beans from Angola were kindly supplied by the International Coffee Organisation, London. The beans were frozen, ground and extracted with 70% MeOH as previously described [2]. The bulked extracts were treated with Carrez Reagent (1 ml A plus 1 ml B) to ppt. colloidal material, filtered and evapd to dryness at red. pres. The residue was redissolved in a minimal vol. of MeOH, refiltered and used for prep. HPLC. Caffeic acid and *L*-tyrosine were obtained from Sigma. All other reagents were normal commercial items of good quality.

\*Current Address: Croner Publications Ltd, Croner House, 173, Kingston Road, New Malden, Surrey KT3 3SS, U.K.

Table 1.  $^1\text{H}$  NMR spectral data for component 7, L-tyrosine and caffeic acid

| H                                 | Component 7                                       | L-Tyrosine                                        | Caffeic acid                          |
|-----------------------------------|---------------------------------------------------|---------------------------------------------------|---------------------------------------|
| H <sub>2</sub> and H <sub>6</sub> | 6.79, 2H, <i>d</i> , <i>J</i> = 8 Hz              | 6.84, 2H, <i>d</i> , <i>J</i> = 8 Hz              |                                       |
| H <sub>3</sub> and H <sub>5</sub> | 7.12, 2H, <i>d</i> , <i>J</i> = 8 Hz              | 7.18, 2H, <i>d</i> , <i>J</i> = 8 Hz              |                                       |
| H $\alpha$                        | { 4.60, 1H, <i>t</i> ,<br><i>J</i> = 3 and 9 Hz   | { 3.82, 1H, <i>t</i> ,<br><i>J</i> = 3 and 9 Hz   |                                       |
| H $\beta$ , ax                    | { 2.94, 1H, <i>dd</i> ,<br><i>J</i> = 9 and 15 Hz | { 3.01, 1H, <i>dd</i> ,<br><i>J</i> = 9 and 15 Hz |                                       |
| H $\beta$ , eq                    | { 3.15, 1H, <i>dd</i> ,<br><i>J</i> = 15 and 5 Hz | { 3.21, 1H, <i>dd</i> ,<br><i>J</i> = 15 and 5 Hz |                                       |
| H $\alpha'$                       | 6.35, 1H, <i>d</i> , <i>J</i> = 16 Hz             |                                                   | 6.21, 1H, <i>d</i> , <i>J</i> = 16 Hz |
| H $\beta'$                        | 7.27, 1H, <i>d</i> , <i>J</i> = 16 Hz             |                                                   | 7.56, 1H, <i>d</i> , <i>J</i> = 16 Hz |
| H <sub>2'</sub>                   | 7.05, 1H, <i>s</i>                                |                                                   | 7.02, 1H, <i>s</i>                    |
| H <sub>5'</sub>                   | 6.97, 1H, <i>d</i> , <i>J</i> = 8 Hz              |                                                   | 6.93, 1H, <i>d</i> , <i>J</i> = 8 Hz  |
| H <sub>6'</sub>                   | 6.86, 1H, <i>d</i> , <i>J</i> = 8 Hz              |                                                   | 6.78, 1H, <i>d</i> , <i>J</i> = 8 Hz  |

**General.** Analytical HPLC of CGA-like components was performed as previously described [3] using a 3  $\mu\text{m}$  reversed phase non-end capped C<sub>18</sub> packing and an acidic (pH 2.5) acetonitrile gradient. The eluate was monitored sequentially at 280 and 313 nm. Prep. HPLC was performed in a similar manner but using a 25 cm  $\times$  8 mm column containing the equivalent 5  $\mu\text{m}$  packing and a non-linear gradient profile [4]. Analytical amino acid analysis was performed on a Waters Pico Tag<sup>TM</sup> chromatographic system using a reversed phase column marketed for the analysis of protein hydrolysates and free amino acids after PITC derivatisation.

$^1\text{H}$  NMR spectra were obtained at 300 MHz. Samples, dissolved in D<sub>2</sub>O or D<sub>2</sub>O-CD<sub>3</sub>OD mixtures as appropriate, were examined at room temp. against a TMS standard.

A UV spectrum was obtained in 70% MeOH, against a 70% MeOH blank.

**Acid hydrolysis.** An aliquot (100  $\mu\text{l}$ ) of component 7 was refluxed with 6 M HCl (3 ml) and the hydrolysate sampled at 1, 4 and 7 hr. The hydrolysate was used directly for phenols analysis, and after derivatisation with PITC for amino acid analysis.

**Transesterification.** An aliquot (100  $\mu\text{l}$ ) of component 7 was treated with 10  $\mu\text{l}$  tetramethylammoniumhydroxide (20% in EtOH) for 5 min at room temp. The reaction was stopped by adding 20  $\mu\text{l}$  2.5 M HOAc, and the reaction mixture analysed for phenols by direct injection.

**Spot tests.** Aliquots (10  $\mu\text{l}$ ) of component 7 were spotted on

Whatman No 1 filter papers and treated individually with the ninhydrin reagent, the 4-dimethylaminocinnamaldehyde reagent [7] or the molybdate reagent [8]. The ninhydrin treated specimen was heated at 100° for 10 min before examination. Reagent blanks and positive controls (tyrosine, tryptophan and caffeic acid) were performed simultaneously.

**Acknowledgements**—This investigation was in part supported by AFRC Grant 90/25, and partly by IESTE. Fraulein Jutta Blank and Fraulein Andrea Henne are thanked for technical assistance. The  $^1\text{H}$  NMR was performed by Mr J. Bloxsidge of the Chemistry Department.

#### REFERENCES

1. Pictet, G. and Brandenberger, H. (1950) *J. Chromatogr.* **4**, 396.
2. Clifford, M. N. (1986) *Phytochemistry* **25**, 1767.
3. Clifford, M. N. and Jarvis, T. (1988) *Food Chem.* **29**, 291.
4. Clifford, M. N., Kellard, B. and Birch, G. G. (1989) *Food Chem.* **33**, 115.
5. Morishita, H., Takai, Y., Yamada, H., Fukada, F., Savada, M., Iwahashi, H. and Kido, R. (1987) *Phytochemistry* **26**, 1195.
6. Meilgaard, P. and Ravn, H. (1988) *Phytochemistry* **27**, 2411.
7. Harborne, J. B. (1985) *Phytochemical Methods*. Chapman & Hall.
8. Clifford, M. N. and Wight, J. (1976) *J. Sci. Food Agric.* **27**, 73.